

Radial velocity and line broadening in spectra

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SMSW2, 22 February 2019



Background

Physics
Carleton College (Northfield MN)

) Astronomy

University of Wisconsin (Madison)
Early conditions of massive star
forming environments
Internal properties of infrared
Blark clouds

College of Idaho
Physics professor since 2009
ntro and upper division physics
ntro astronomy

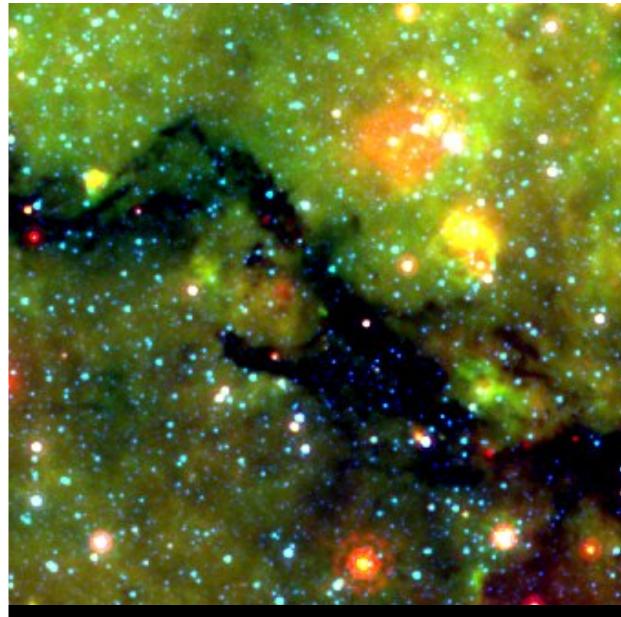


Image credit: Milky Way Project (milkywayproject.org)
Infrared dark clouds are seen in silhouette aga
background glowing dust emission.

Research

ssive Star Formation

Radio: VLA, Green Bank Telescope

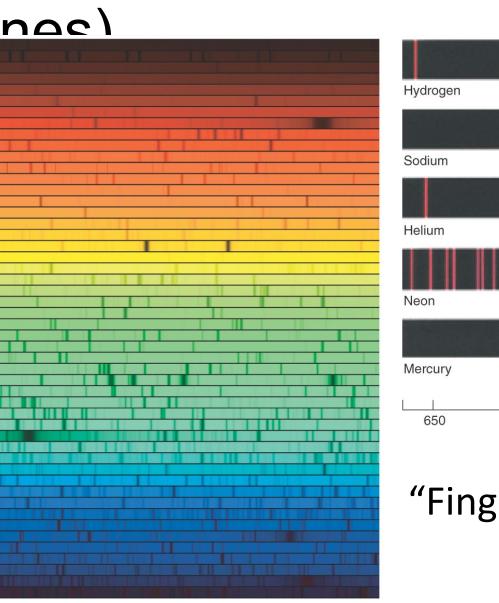
nfrared: Spitzer, Herschel

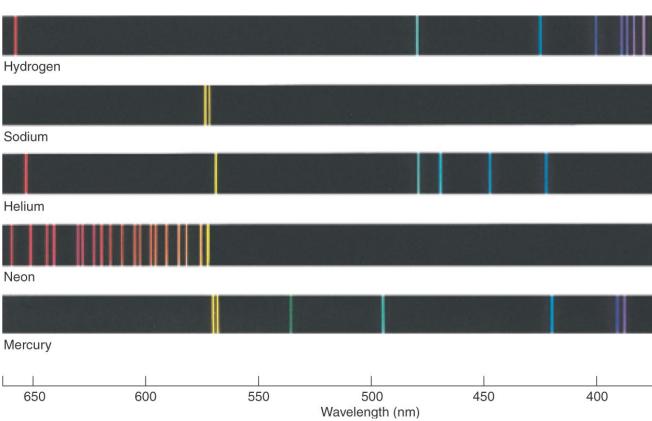
rarchical star formation Bubbles Yellowballs

connection to this ference: I use radio spectral s to determine conditions of ecular gas in star forming as.



mission and Absorption Lines (Spectral





"Fingerprints" of atoms and molecule

Images: Pearson e

Atoms and Absorption Lines: The usual model

Absorption can boost an electron to the second (or higher) excited state

Ways to decay:

- 1. To ground state
- 2. Cascade one orbital at a time

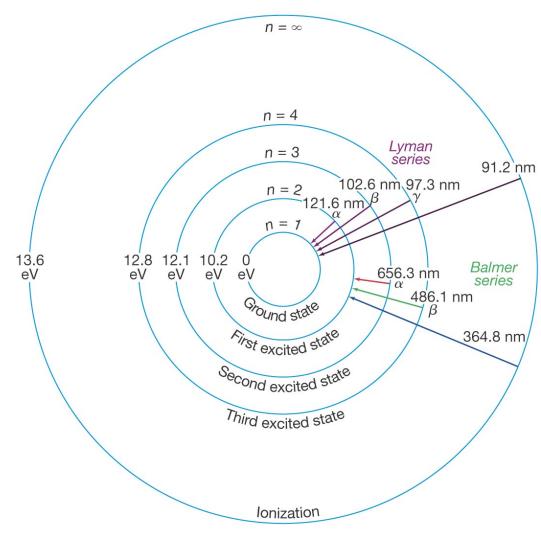


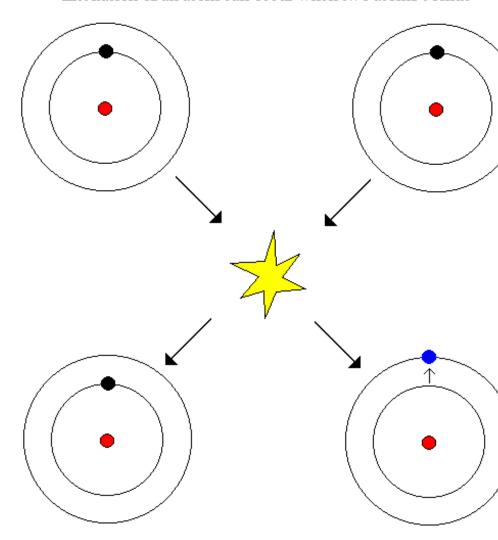
Image: Pearson ed

nere are other ways generate emission es besides "eating notons"

isional excitation: can be used to neasure the density, emperature of gas

Collisional Excitation

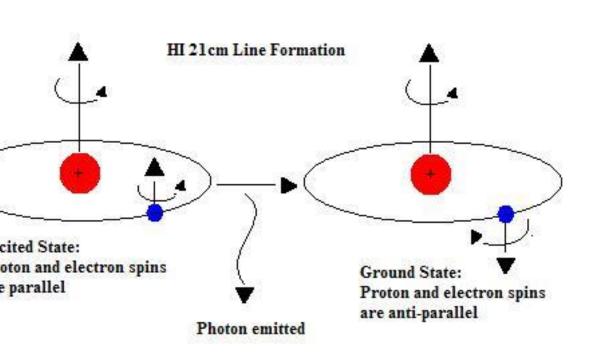
Excitation of an atom can occur when two atoms collide



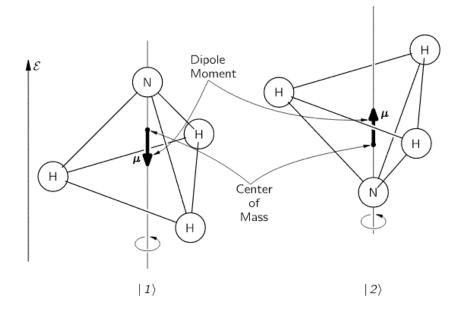
some of the energy of the collision is transferred to the electron bumping it to a higher orbit.

Image: http://abyss.uoregon.edu/~js/glossary/excita

Additional types of energy levels of atoms and molecules

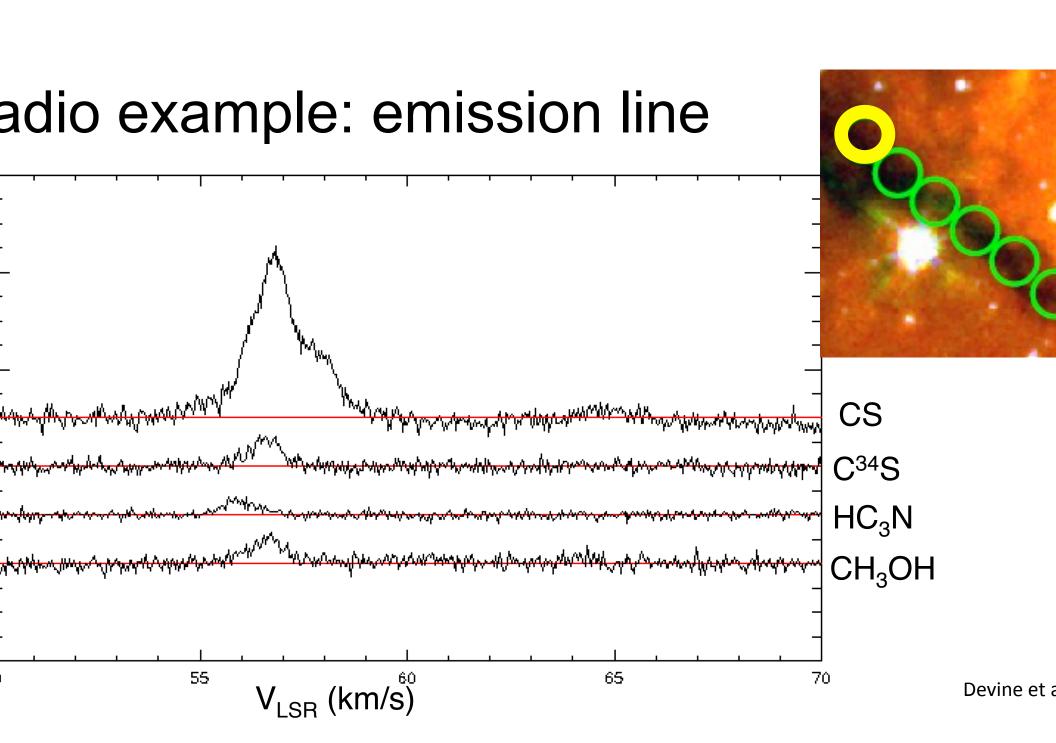


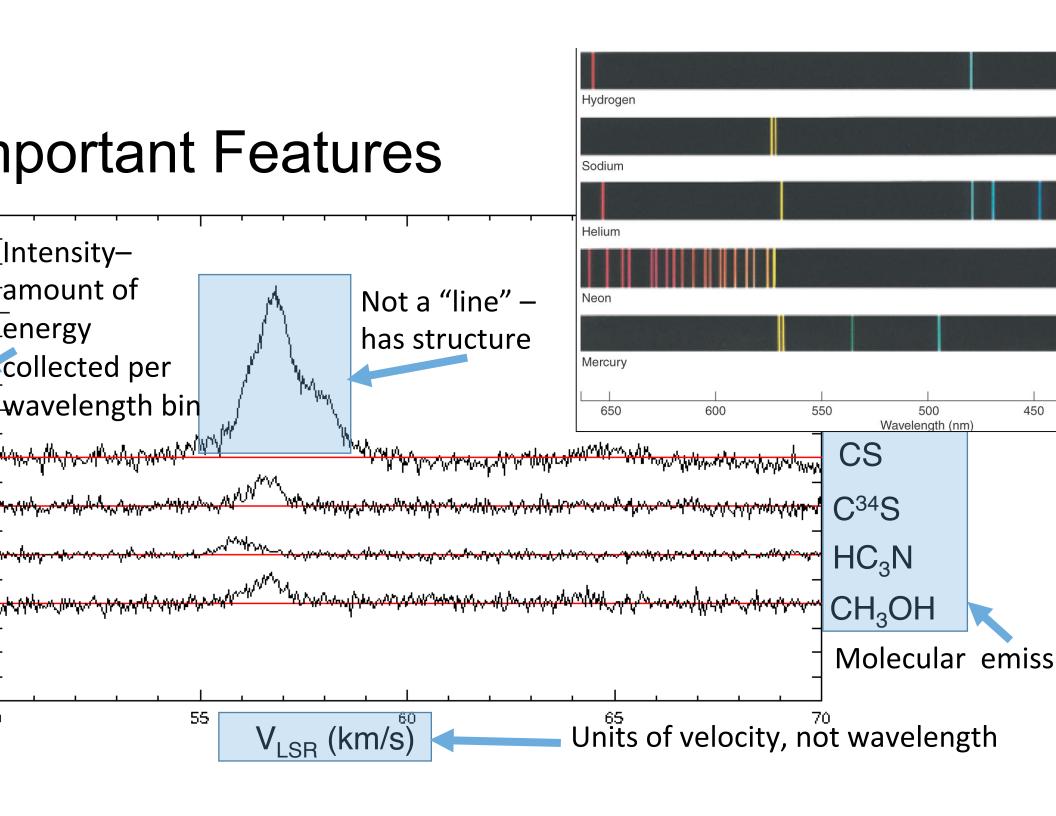
ctron/proton spin alignments have ifferent energies. Example: 21 cm ne in hydrogen.



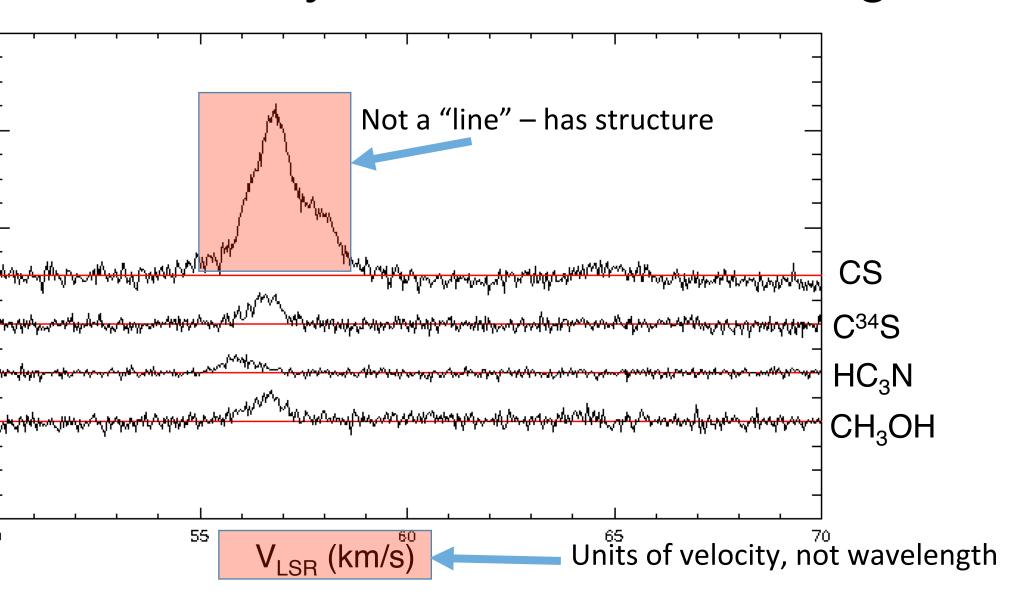
Molecule's direction of rotar around its axis of symmet Example: position of N at ammonia (NH₃).

Image: teachastronomy.com, readingfeynma

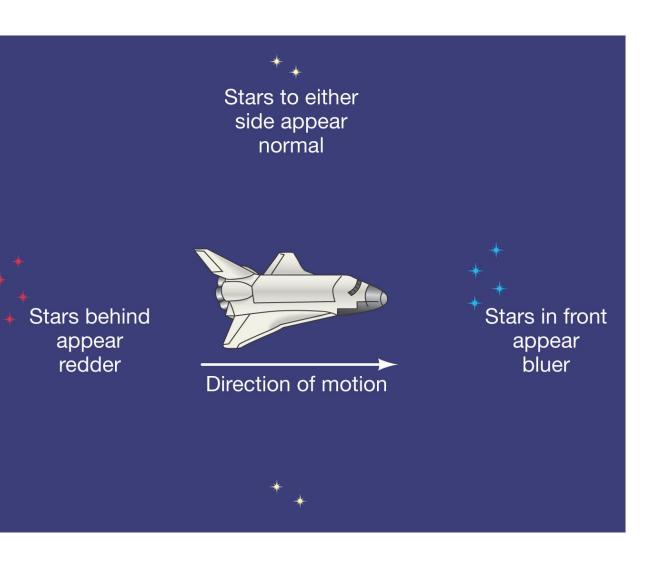




adial Velocity and Line Broadening



The Doppler Effect



Light can't travel faster (o slower) than the speed of light, even if it is emitted from a moving source.

If one is moving toward a source of radiation, the wavelengths seem shorte (higher energy photons); i moving away, they seem longer (lower energy photons).

The Doppler Effect

tion of source and erver... motion must be dial" (towards or away in line of sight) to cause opler Effect

$$=v/c$$

re shift = higher radial ocity

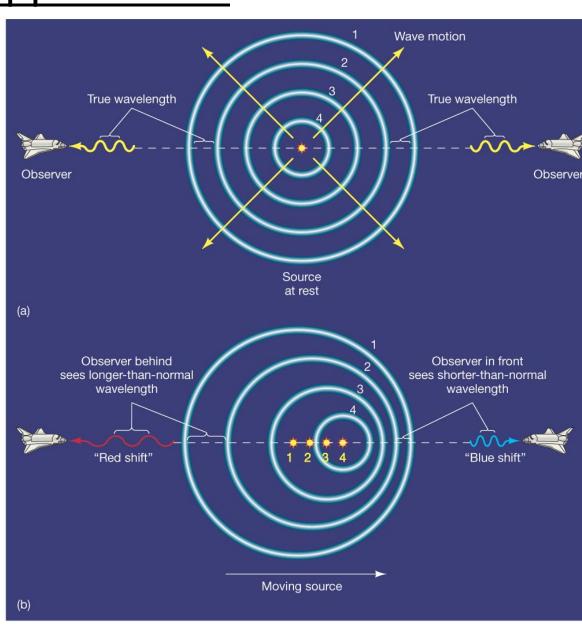
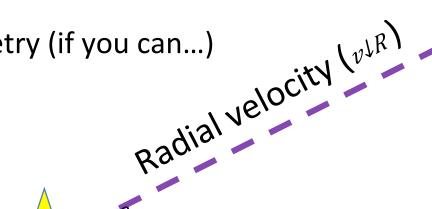


Image: Pearson ed

Radial velocity vs. source velocity

adial velocity: use Doppler Shift

ransverse velocity: astrometry (if you can...)



Source velocity (vls)

$$v \downarrow R = v \downarrow s \cos \Box \theta$$

$$v \downarrow S \uparrow 2 = v \downarrow R \uparrow 2 + v \downarrow T \uparrow 2$$



Velocity in Our Galaxy: Galactic Rotatio

Radial velocity (relative to Sun) of gas in Milky Way spiral arms

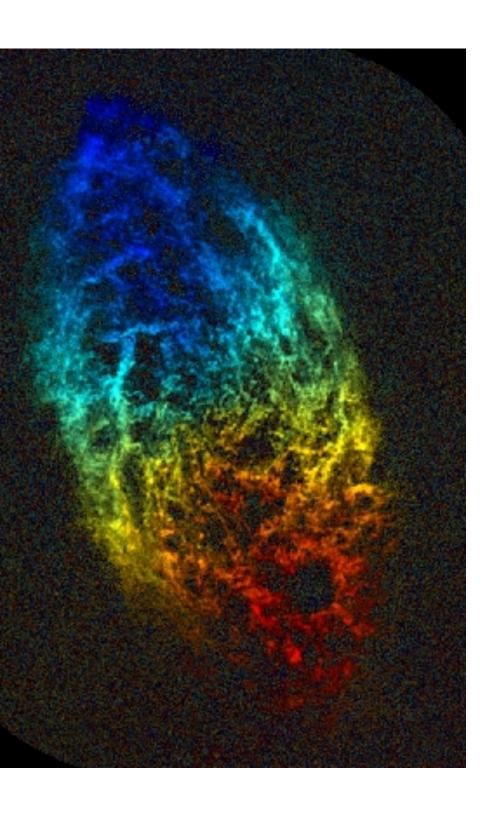
Image: NASA (APOD 25 A 2005)

CrNNor -55 10 10 0 X (kpc)

Velocity in Our Galaxy: Galactic Rotatio

Radial velocity as tool to getting distances to object in spiral arms ("We obtained source distance using a kinematic model...

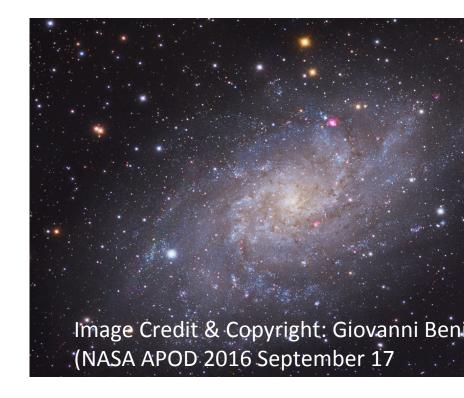
Reid et al., ApJ, 2016



Gas in other galaxies Hydrogen in M33 (Triangulum or

Hydrogen in M33 (Triangulum or Pinwheel Galaxy) imaged by VLA and Westerbork telescopes.

Image: NRAO, https://www.nrao.edu/pr/2001/m33gas/



Relation Between Redshift and Distance for Distant Galaxies uster Distance in Redshifts alaxy in: light years H+K lines 78,000,000 1,200 km.s 1,000,000,000 1,400,000,000 rona Borealis 22,000 km.s 2,500,000,000 otes 3,960,000,000

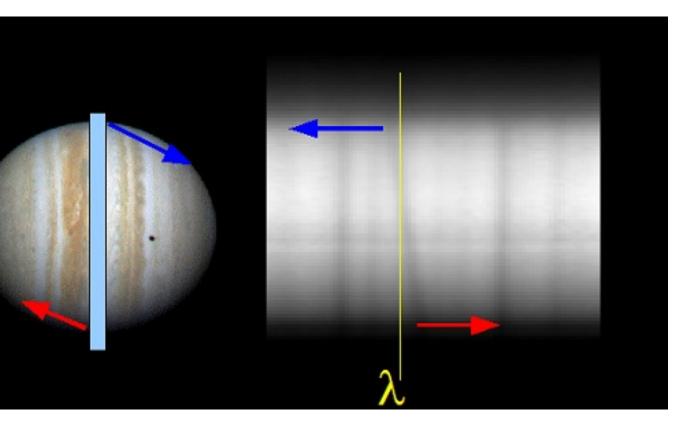
H+K lines

61,000 km.s⁻¹

Velocities of other galaxies

Edwin Hubble's observations of galaxies with the redshift in their spectral lines (1943).

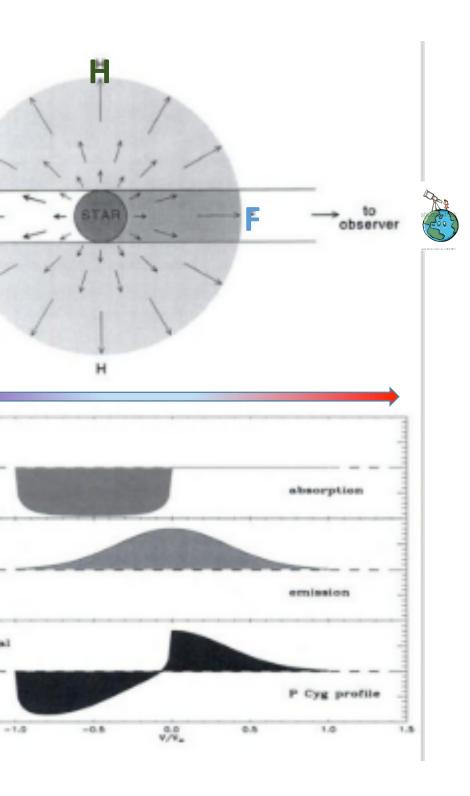
Expansion, motion within Galaxy clusters, rotation galaxies



Planetary Rotation

Example: Rotation of Ju

Image: Shelyak.com

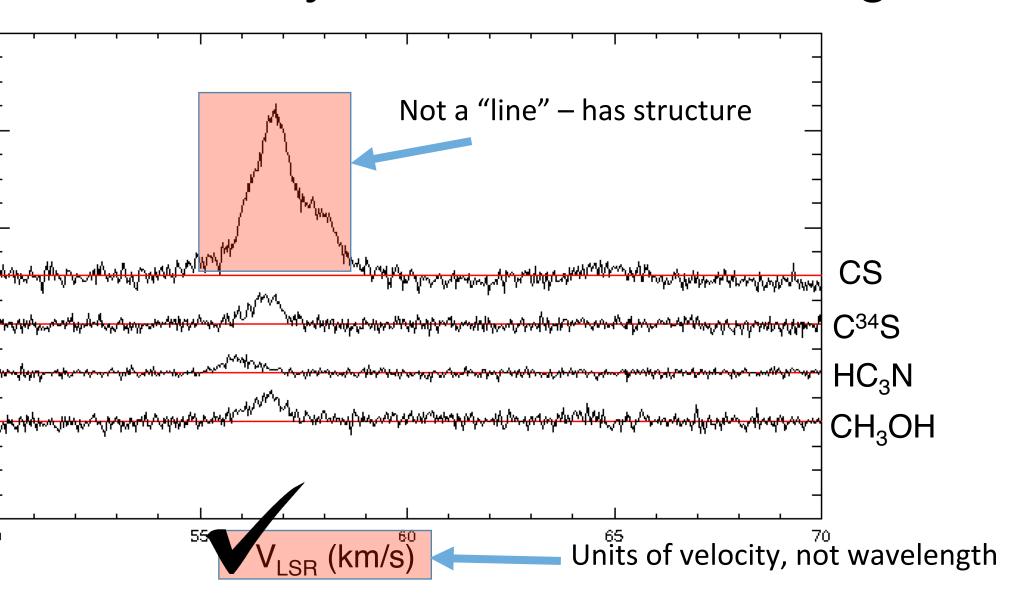


P-Cygni Profiles

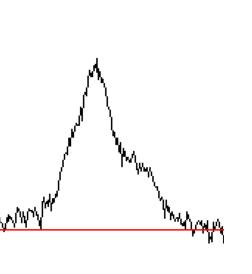
Bottom shows the continuum from star with absorption in front (F) are emission from halo (H)

Image: Figure 2.4 from Lamers & Cassinelli, Introduction to Stellar WI 1999, Cambridge University Press

adial Velocity and Line Broadening



ne Broadening



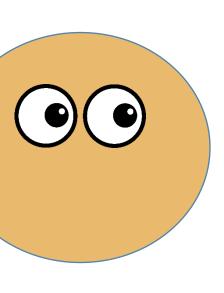
Dispersion:

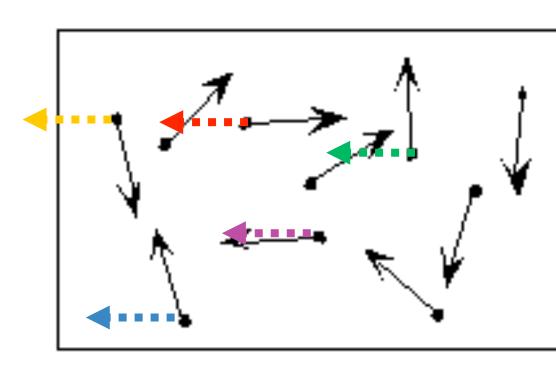
Natural broadening Pressure broadening

Gaussian:

Thermal Turbulence

See David Whelan's talk for more!





(molecular motion)

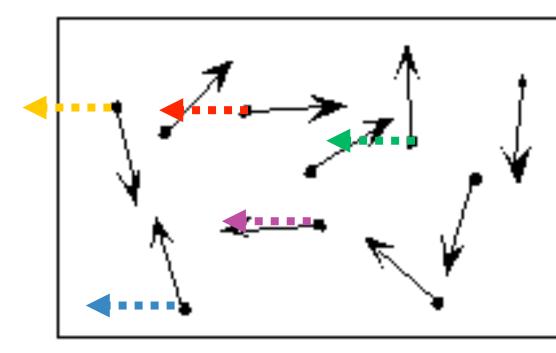
v is fraction of atoms with velocity between v and v+dv

Image credit: http://cronodon.com/SpaceTech/CVAccretic

v)dv is fraction of atoms with tity between v and v+dv

dv is "probability distribution" depends on gas properties

ability distribution determines ne shape



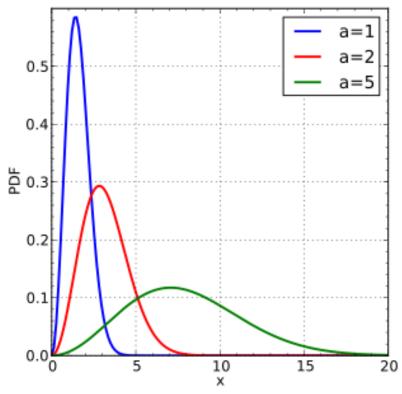
(molecular motion)

Image credit: http://cronodon.com/SpaceTech/CVAccretic

v)dv is fraction of atoms with eity between v and v+dv

 y_{jdv} is "probability distribution" depends on gas properties

ability distribution determines ne shape

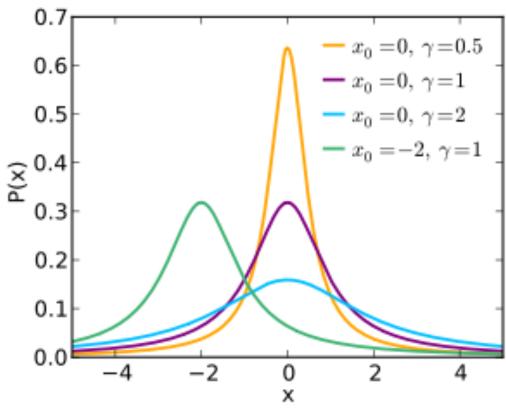


Maxwell-Boltzmann distribution— ideal gwith random motions

v)dv is fraction of atoms with tity between v and v+dv

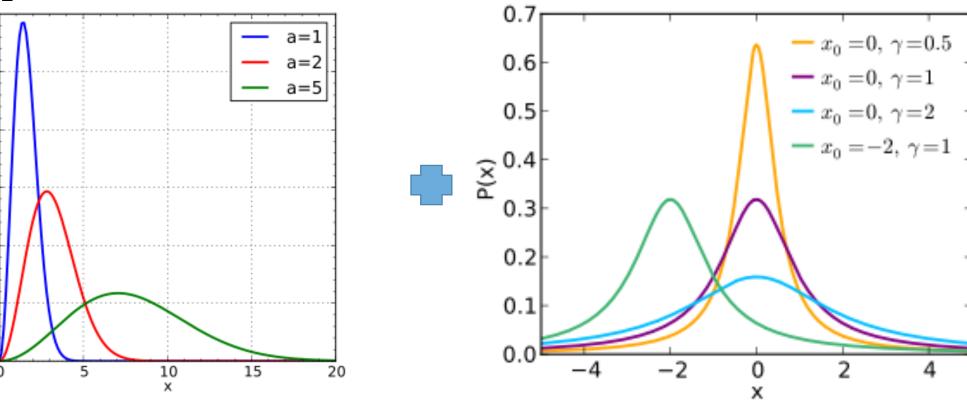
 y_{jdv} is "probability distribution" depends on gas properties

ability distribution determines ne shape



Cauchy-Lorentz (Lorentzian) distribution homogeneous broadening

Image credit: http://cronodon.com/SpaceTech/CVAccretic



axwell-Boltzmann distribution: nermal (Doppler) broadening

Lorentz distribution:
Natural (uncertainty principle) broadeni
Collisional broadening

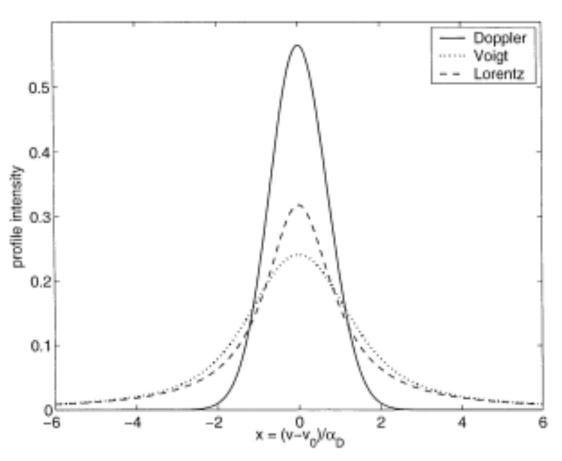
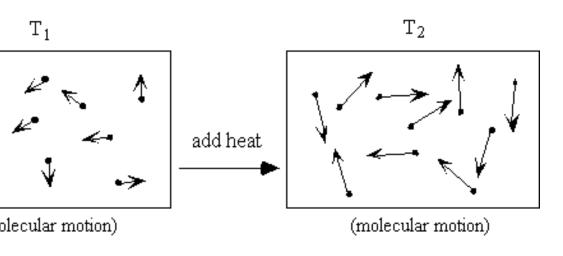


Fig. 2. The Lorentz profile (dashed line) and the Doppler profile (solid line) with the same half-widths ($\alpha_L = \alpha_D = 1$). The dotted line is the corresponding Voigt profile. Here, x is defined as $x \equiv (v - v_0)/\alpha_D$.

Voigt profile: convolution of Gaussian and Lorentzian profiles

Figure: Huang & Yung, "A
Common Misunderstanding
about the Voigt Line Profile" 2003
Journal of The Atmospheric
Sciences

Example: Thermal Motion Temperature is Energy, Energy is Motion



 $T=2/3 KE \downarrow avg /k \downarrow b$

k_h is Boltzmann's constant, 1.38 x10⁻²

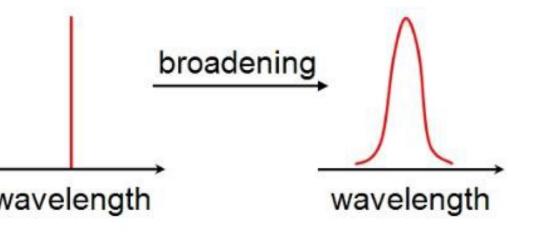
Rearrange to get average kinetic ener of a gas:

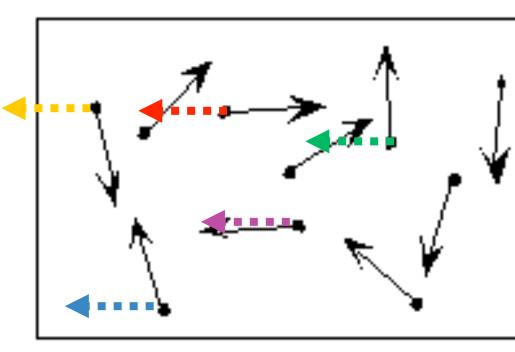
 $KE \downarrow avg = 3/2 \ k \downarrow b \ T = 1/2 \ mv \downarrow average 12$

ermal Linewidths

ning gas random motions follow Maxwellnann distribution:

$$(v)=\sqrt{\Box(m/2\pi kT)}$$
 13 $4\pi v$ 12 $\exp\Box[-mv$ 12 $/2kT]$





(molecular motion)

FWHM=1/λ $\sqrt{\Box}$ 8 kT/m ln \Box 2

Image credit: spectral line broadening via BotReje at http://cronodon.com/SpaceTech/CVAccretion

Velocity in Gas: Turbulent motion



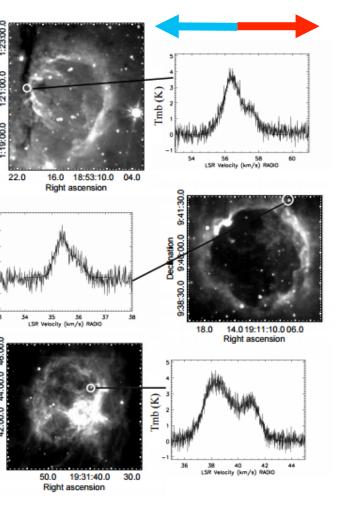
Stars, Gas, and Dust Battle the Carina Nebula

Image Credit & Copyright: Bastien Fouch

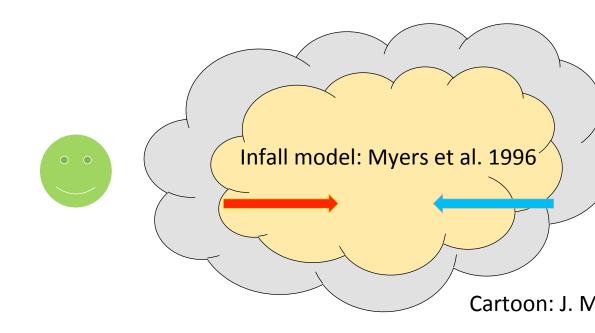
NASA APOD 2017 August 15

Measure the temperature of a gas, and find that lines are broader than thermal linewidths \rightarrow there is probably turbulent motion

Fun science example: using spectra to ook for evidence of in-falling clouds



Watson, Devine et al. 2016: CS lin profiles show evidence of infall



62-1, b) N90-2 and c) N117-3.

n science example: using spectra to amine motion of gas in clouds

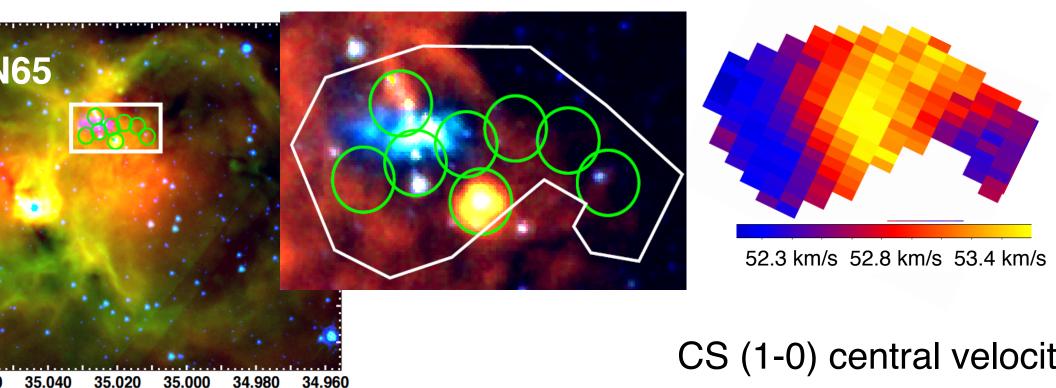


image of N65. White outline shows region apped in CS (1-0).

n science example: using line profiles interpret Be Star spectra

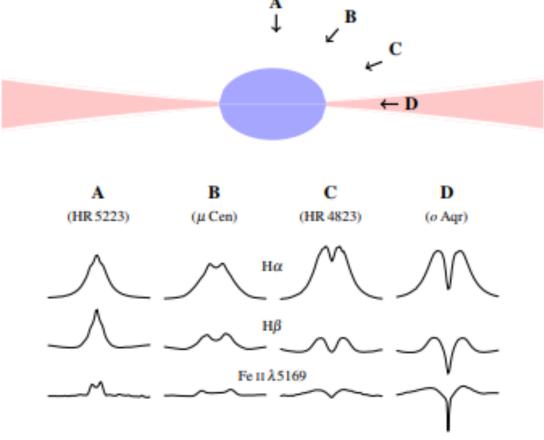


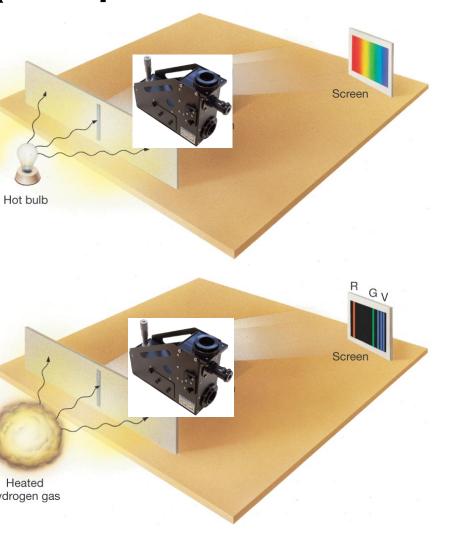
Fig. 1 Schematic view of a Be star at critical rotation and with a flared disk. The lower part shows example spectral profiles from pole-on to shell Be stars

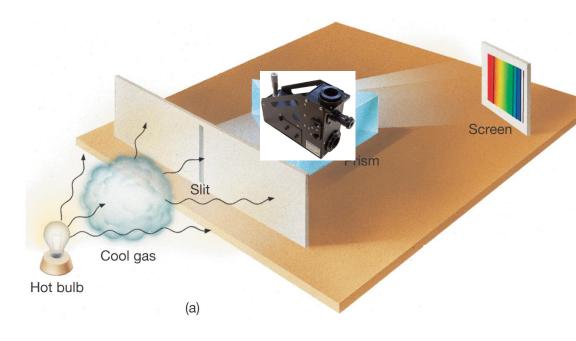
Rivinius, Carciofi & Marta

Summary

- Origin of emission/absorption spectra, conditions for each (Kirchhoff's Laws)
- Doppler shift, and how to convert from wavelength to line-of-sight velocity
- Bulk motions of gas: galactic rotations, galactic motion, planetary rotation
- Motion within gas and line broadening, thermal vs. turbulent motion
- Science examples: using line profiles to probe star formation, dense gas conditions, stellar atmospheres

Review: Sources of Emission (simplified version)





Images: Pearson educ Shelyak Instruments (